## Macroscopic Electrodynamics

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## Previously on CHM676...

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Today: How to simplify the equations by *coarse-graining*.







#### Statistical Mechanics: Average Physics

Microstate

Macrostate



#### Statistical Mechanics: Average Physics



#### Ensemble Averages

Formally, we define the ensemble average

$$\langle f \rangle_M = \sum_{\mu \in M} p_\mu^{(M)} f(\mu)$$

as the average value over all microstates consistent with a specific macrostate (temperature, volume, charge, etc.).

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In spectroscopy, we work almost exclusively with the *macroscopic fields* 

$$\boldsymbol{E}(\boldsymbol{x},t) \equiv \langle \boldsymbol{e}(\boldsymbol{x},t) \rangle_{M} \boldsymbol{B}(\boldsymbol{x},t) \equiv \langle \boldsymbol{b}(\boldsymbol{x},t) \rangle_{M} .$$

#### Macroscopic Fields

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## Macroscopic Fields

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Since Maxwell's equations are linear in b and e:

$$\nabla \cdot \boldsymbol{E} = 4\pi \langle \varrho(\boldsymbol{x}, t) \rangle_M$$
$$\nabla \cdot \boldsymbol{B} = 0$$
$$\nabla \times \boldsymbol{E} + \frac{1}{c} \frac{\partial \boldsymbol{B}}{\partial t} = 0$$
$$\nabla \times \boldsymbol{B} - \frac{1}{c} \frac{\partial \boldsymbol{E}}{\partial t} = \frac{4\pi}{c} \langle \boldsymbol{j}(\boldsymbol{x}, t) \rangle_M$$

But how do we get rid of the explicit ensemble average? *Wait for it...* 

### **Take-Home Points**

In statistical mechanics, an *ensemble-averaged* quantity is the average value over all *microstates* consistent with a given *macrostate*.

In spectroscopy, we primarily deal with the *macroscopic* fields E and B – the ensemble averages of the *microsocpic* fields e and b.

Since Maxwell's equations are linear, their form is unchanged by ensemble-averaging. The key question is how to handle  $\langle \varrho \rangle$  and  $\langle j \rangle$ .

# Coarse-grained Densities

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A two-step process:

(1) Distinguish between *free* and *bound* charges:

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(2) Multipole expansion in molecular size

- Monopole: Total charge
- Oipole: Polarization

https://en.wikipedia.org/wiki/File: Ball\_and\_stick\_model\_of\_a\_water\_molecule.png





## Macroscopic Densities

The result (after great suffering) is:

$$\begin{split} \langle \varrho(\boldsymbol{x},t)\rangle_{M} &\approx \rho(\boldsymbol{x},t) - \nabla \cdot \boldsymbol{P}(\boldsymbol{x},t) + \dots \\ \langle \boldsymbol{j}(\boldsymbol{x},t)\rangle_{M} &\approx \boldsymbol{J}(\boldsymbol{x},t) + \frac{\partial \boldsymbol{P}(\boldsymbol{x},t)}{\partial t} + \dots \end{split}$$

where

- $\rho$  is the *free charge* density
- **P** is the *polarization* density
- J is the *free current* density

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Coarse-grained Densities

## Microscopic Charge Density: $\varrho$

## How much total charge?



## Free Charge Density: $\rho$

### How much mobile charge?



#### Free Current Density: J

### How much mobile charge movement?



#### Polarization Density: P

## How much polarization?



 $\begin{array}{l} \boldsymbol{P}(\boldsymbol{x},t) \text{ is the average molecular dipole at } \boldsymbol{x} \text{:} \\ \boldsymbol{P}(\boldsymbol{x},t) \equiv \left\langle \mu^{\mathsf{mol}} \delta(\boldsymbol{x}-\boldsymbol{r}^{\mathsf{mol}}) \right\rangle_{M} \end{array}$ 

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## Read the Signs

	Microscopic	Macroscopic
Electric Field	e	E
Magnetic Field	b	B
Charge Density	Q	$\rho$
Current Density	j	J

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$$\begin{split} \langle \varrho(\boldsymbol{x},t) \rangle_M &\approx \rho(\boldsymbol{x},t) - \nabla \cdot \boldsymbol{P}(\boldsymbol{x},t) + \dots \\ \langle \boldsymbol{j}(\boldsymbol{x},t) \rangle_M &\approx \boldsymbol{J}(\boldsymbol{x},t) + \frac{\partial \boldsymbol{P}(\boldsymbol{x},t)}{\partial t} + \dots \end{split}$$

Free charges can move over large distances

Bound charges are mostly stuck in place

The multipole expansion allows us to write  $\langle \varrho \rangle_M$  and  $\langle j \rangle_M$  in terms of *macroscopic quantities:* 

- The free charge density  $\rho$
- ullet The polarization density  $oldsymbol{P}$
- ullet The free current density  $oldsymbol{J}$

## Molecular Spectroscopy and Material Polarization

#### Going Forward: Dielectric Materials

In this course we'll work with *dielectrics*: materials without free charges. *Whole molecules don't move very fast!* 

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In this course we'll work with *dielectrics*: materials without free charges. *Whole molecules don't move very fast!* 

In dielectrics, both  $\rho$  and  $\boldsymbol{J}$  vanish:

$$\nabla \cdot \boldsymbol{E} = -4\pi \nabla \cdot \boldsymbol{P}(\boldsymbol{x}, t)$$
$$\nabla \cdot \boldsymbol{B} = 0$$
$$\nabla \times \boldsymbol{E} + \frac{1}{c} \frac{\partial \boldsymbol{B}}{\partial t} = 0$$
$$\nabla \times \boldsymbol{B} - \frac{1}{c} \frac{\partial \boldsymbol{E}}{\partial t} = \frac{4\pi}{c} \frac{\partial \boldsymbol{P}(\boldsymbol{x}, t)}{\partial t}$$

For molecular spectroscopy, the *polarization density* is key.

## Think about it!

**Q:** How does polarization contribute to  $\langle \varrho \rangle_M$ ?



## Think about it!

**Q:** How does polarization contribute to  $\langle \varrho \rangle_M$ ?



A: It doesn't! It's  $\nabla \cdot P!$ 

Molecular Spectroscopy and Material Polarization

#### Next up: Response Theory



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Molecular Spectroscopy and Material Polarization

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#### **Take-Home Points**

P doesn't contribute directly to  $\langle \varrho \rangle_M$  or  $\langle j \rangle_M$ : It's  $\nabla \cdot P$  and  $\frac{\partial P}{\partial t}$  that contribute.

In macroscopic electrodynamics, *response theory* replaces the Lorentz force law in describing how materials respond to the EM field.